

APPENDIX 11 D
PROCESS TRAILER CARBON FILTRATION UNIT AND UNPACK AREA CARBON
FILTER UNIT CARBON BREAKTHROUGH CALCULATION SHEETS

CALCULATION SHEET

Calculation # _____

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Revision # _____

Attachments YES / NO

Calculation Sheet MMD-1 UNPACK AREA CARBON FILTER DESIGNPrepared by C. CUDWORTH Date 12/13/96

Checked by _____ Date _____

References Documents:

Project # _____

PURPOSE : TO DETERMINE THE BREAK THROUGH TIME OF THE 1ST 2" THICK CARBON BED OF THE UNPACK AREA CARBON FILTER UNIT

EQUATIONS :

$$t_b = \frac{P_b W_e}{V_L C_o} [\lambda - \lambda_c] \quad \text{WHERE:}$$

$$\lambda_c = \frac{V_L}{K_v} \ln \frac{C_o}{C_b}$$

$$K_v = 1.82 a V_L \left[\frac{D_p G}{\mu} \right]^{-0.51} \left[\frac{\mu}{P D_v} \right]^{-2/3}$$

t_b = TIME TO BREAKTHROUGH, S
 C_o = CHALLENGE CONCENTRATION OF AGENT, g/cm³
 W_e = CARBON CAPACITY, g/g
 P_b = BULK (APPARENT) DENSITY OF CARBON, g/cm³
 λ = DEPTH OF CARBON BED, CM
 λ_c = CRITICAL BED DEPTH, CM
 V_L = LINEAR VELOCITY OF AIR THROUGH BED, CM/S
 C_b = BREAKTHROUGH CONCENTRATION OF AGENT, g/cm³
 K_v = FIRST ORDER RATE CONSTANT FOR ADSORPTION, S
 a = PARTICLE SURFACE AREA / BED VOLUME $(1-E)/D_p$, CM⁻¹
 D_p = PARTICLE DIAMETER, CM
 E = BED VOID FRACTION
 G = MASS FLOW RATE OF AIR ($P V_L$), g/CM²-S
 μ = VISCOSITY OF AIR, g/CM-S
 P = DENSITY OF AIR, g/CM³
 D_v = DIFFUSIVITY OF AGENT IN AIR, CM²/S
 (ABOUT 0.07 CM²/S FOR GB AT AMBIENT)

INPUT DATA :

- CHALLENGE CONCENTRATION IS BASED ON JACADS DESIGN BASIS MAXIMUM FILTER CHALLENGE OF 40 mg/m³ CONTAINED IN MIRE REPORT #93W0000034, ASSESSMENT OF CARBON FILTER PERFORMANCE, 9/93
- GB AS THE MOST VOLATILE AND LEAST FAVORABLY ADSORBED AGENT IS A SUITABLE SURROGATE FOR THE OTHER AGENTS MUSTARD AND VX, WHICH ARE CAPTURED BY THE MECHANISM OF MECHANICAL ADSORPTION.

$C_o = 40 \text{ mg/m}^3 = 4 \times 10^{-8} \text{ g/cm}^3$ CHALLENGE CONCENTRATION OF GB
 $P_b = 0.63 \text{ g/cm}^3$ MINIMUM BULK APPARENT DENSITY - SPEC
 $D_p = 0.10 \text{ cm}$ AVERAGE PARTICLE SIZE 12 X 30 MESH
 $W_e = 0.21 \text{ g/g}$ CARBON CAPACITY FOR GB
 $V_L = 14.23 \text{ cm/s}$
 $\lambda = 2" = 5.08 \text{ cm}$
 $C_b = 0.0001 \text{ mg/m}^3 = 1 \times 10^{-13} \text{ g/cm}^3$ FROM MMD SPEC FOR G2
 $D_v = 0.07 \text{ cm}^2/\text{s}$
 $\mu = 187.499 \times 10^{-6} \text{ g/cm-s}$ VISCOSITY OF AIR @ 68°F
 $P = 1.1833 \times 10^{-3} \text{ g/cm}^3$ DENSITY OF AIR

CALCULATION SHEET

Calculation # _____

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Attachments YES / NO

Calculation Sheet *MMD-1 UNPACK AREA CARBON FILTER DESIGN*

Prepared by _____ Date _____

Checked by _____ Date _____

References Documents:

Project # _____

$$\epsilon = 0.4$$

$$a = 6(1-\epsilon)/D_p = 6(1-0.4)/0.10 = 36 \text{ cm}^{-1}$$

$$G = P V_L = (1.1833 \times 10^{-3})(15.4) = 18.223 \times 10^{-3} \text{ g/cm}^2 \cdot \text{s}$$

$$K_v = (1.82)(36)(14.23)(9.719)^{-0.51} (2.264)^{-2/3}$$

$$K_v = (1.82)(36)(14.23)(.3136)(.5799)$$

$$K_v = 169.55 \text{ s}^{-1}$$

$$\lambda_c = \frac{14.23}{169.55} \ln \frac{4 \times 10^{-8}}{1 \times 10^{-13}}$$

$$= 1.08 \text{ cm}$$

$$t_b = (0.63)(0.21)(5.68 - 1.08) / (14.23)(4 \times 10^{-8})$$

$$t_b = 929,700 \text{ SEC} = 258 \text{ HR}$$

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Attachments YES / NO

Calculation Sheet MMD-1 PROCESS TRAILER CARBON FILTER DESIGN

Prepared by C. CUDWORTH Date 12/13/96

Checked by _____ Date _____

References Documents:

Project # _____

PURPOSE: TO DETERMINE THE BREAK THROUGH TIME OF THE 1ST 2" THICK CARBON BED OF THE PROCESS TRAILER CARBON FILTER UNIT.

EQUATIONS:

$$t_b = \frac{P_b W_e}{V_L C_o} [\lambda - \lambda_c]$$

WHERE:

t_b = TIME TO BREAKTHROUGH, S
 C_o = CHALLENGE CONCENTRATION OF AGENT, g/cm³
 W_e = CARBON CAPACITY, g/g
 P_b = BULK (APPARENT) DENSITY OF CARBON, g/cm³
 λ = DEPTH OF CARBON BED, CM
 λ_c = CRITICAL BED DEPTH, CM
 V_L = LINEAR VELOCITY OF AIR THROUGH BED, CM/S
 C_b = BREAKTHROUGH CONCENTRATION OF AGENT, g/cm³
 K_v = FIRST ORDER RATE CONSTANT FOR ADSORPTION, S⁻¹
 a = PARTICLE SURFACE AREA / BED VOLUME
 $\frac{6(1-E)}{D_p}$, CM⁻¹
 D_p = PARTICLE DIAMETER, CM
 E = BED VOID FRACTION
 G = MASS FLOW RATE OF AIR (PVL), g/cm²-S
 μ = VISCOSITY OF AIR, g/cm-S
 P = DENSITY OF AIR, g/cm³
 D_v = DIFFUSIVITY OF AGENT IN AIR, CM²/S
 (ABOUT 0.07 CM²/S FOR GB AT AMBIENT)

$$\lambda_c = \frac{V_L}{K_v} \ln \frac{C_o}{C_b}$$

$$K_v = 1.82 a V_L \left[\frac{D_p G}{\mu} \right]^{-0.51} \left[\frac{\mu}{P D_v} \right]^{-2/3}$$

INPUT DATA:

1. CHALLENGE CONCENTRATION IS BASED ON JACADS DESIGN BASIS MAXIMUM FILTER CHALLENGE OF 40 mg/m³ CONTAINED IN MITRE REPORT #93W0000034, ASSESSMENT OF CARBON FILTER PERFORMANCE, 9/93
2. GB AS THE MOST VOLATILE AND LEAST FAVORABLY ADSORBED AGENT IS A SUITABLE SURROGATE FOR THE OTHER AGENTS MUSTARD AND VX, WHICH ARE CAPTURED BY THE MECHANISM OF MECHANICAL ADSORPTION.

$C_o = 40 \text{ mg/m}^3 = 4 \times 10^{-8} \text{ g/cm}^3$ CHALLENGE CONCENTRATION OF GB
 $P_b = 0.63 \text{ g/cm}^3$ MINIMUM BULK APPARENT DENSITY - SPEC
 $D_p = 0.10 \text{ cm}$ AVERAGE PARTICLE SIZE 12X30 MESH
 $W_e = 0.21 \text{ g/g}$ CARBON CAPACITY FOR GB
 $V_L = 15.4 \text{ cm/s}$
 $\lambda = 2" = 5.08 \text{ cm}$
 $C_b = 0.0001 \text{ mg/m}^3 = 1 \times 10^{-13} \text{ g/cm}^3$ FROM MMD SPEC FOR GB
 $D_v = 0.07 \text{ cm}^2/\text{s}$
 $\mu = 187.499 \times 10^{-6} \text{ g/cm-S}$ VISCOSITY OF AIR @ 68°F
 $P = 1.1833 \times 10^{-3} \text{ g/cm}^3$ DENSITY OF AIR

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References Documents:

Project # _____

$$\epsilon = 0.4$$

$$a = \frac{6(1-\epsilon)}{D_p} = \frac{6(1-0.4)}{0.10} = 36 \text{ cm}^{-1}$$

$$G = P_{VL} = (1.1833 \times 10^{-3})(15.4) = 18.223 \times 10^{-3} \text{ g/cm}^2 \cdot \text{s}$$

$$K_v = (1.82)(36)(15.4)(9.719)^{-0.51}(2.264)^{-2/3}$$

$$K_v = (1.82)(36)(15.4)(1.3136)(.5799)$$

$$K_v = 183.52 \text{ s}^{-1}$$

$$\lambda_c = \frac{15.4}{183.52} \ln \frac{4 \times 10^{-8}}{1 \times 10^{-13}}$$

$$\lambda_c = 1.08 \text{ cm}$$

$$t_b = (0.63)(0.21)(5.02 - 1.08) / (15.4)(4 \times 10^{-8})$$

$$t_b = 859,100 \text{ SEC} = 238 \text{ HR}$$